Table of Affibient Air Quality Standards Exhibit i							
Ambient Air Quality Standards							
Pollutant	Averaging Time	California Standards ¹		National Standards ²			
		Concentration ³	Method ⁴	Primary ^{3,5}	Secondary ^{3,6}	Method ⁷	
Ozone (O ₃) ⁸	1 Hour	0.09 ppm (180 μg/m³)	Ultraviolet Photometry	_	Same as Primary Standard	Ultraviolet Photometry	
	8 Hour	0.070 ppm (137 μg/m ³)		0.070 ppm (137 μg/m³)			
Respirable Particulate Matter (PM10) ⁹	24 Hour	50 μg/m ³	Gravimetric or Beta Attenuation	150 μg/m³	Same as Primary Standard	Inertial Separation and Gravimetric Analysis	
	Annual Arithmetic Mean	20 μg/m³		_			
Fine Particulate	24 Hour	_	_	35 μg/m³	Same as Primary Standard	Inertial Separation and Gravimetric Analysis	
Matter (PM2.5) ⁹	Annual Arithmetic Mean	12 μg/m³	Gravimetric or Beta Attenuation	9.0 µg/m³	15.0 µg/m ³		
Carbon Monoxide (CO)	1 Hour	20 ppm (23 mg/m ³)	Non-Dispersive Infrared Photometry (NDIR)	35 ppm (40 mg/m ³)	_	Non-Dispersive Infrared Photometry (NDIR)	
	8 Hour	9.0 ppm (10 mg/m ³)		9 ppm (10 mg/m ³)	_		
	8 Hour (Lake Tahoe)	6 ppm (7 mg/m ³)		_	_		
Nitrogen Dioxide (NO ₂) ¹⁰	1 Hour	0.18 ppm (339 μg/m³)	Gas Phase Chemiluminescence	100 ppb (188 µg/m³)	_	Gas Phase Chemiluminescence	
	Annual Arithmetic Mean	0.030 ppm (57 μg/m ³)		53 ppb (100 μg/m³)	Same as Che Primary Standard		
Sulfur Dioxide (SO ₂) ¹¹	1 Hour	0.25 ppm (655 μg/m ³)	Ultraviolet Fluorescence	75 ppb (196 μg/m³)	_	Ultraviolet Flourescence; Spectrophotometry (Pararosaniline Method)	
	3 Hour			_	0.5 ppm (1300 μg/m³)		
	24 Hour	0.04 ppm (105 μg/m³)		0.14 ppm (for certain areas) ¹¹	_		
	Annual Arithmetic Mean	_		0.030 ppm (for certain areas) ¹¹	_		
Lead ^{12,13}	30 Day Average	1.5 μg/m³	Atomic Absorption	_	_	High Volume Sampler and Atomic Absorption	
	Calendar Quarter	-		1.5 µg/m³ (for certain areas) ¹²	Same as Primary Standard		
	Rolling 3-Month Average	_		0.15 μg/m ³			
Visibility Reducing Particles ¹⁴	8 Hour	See footnote 14	Beta Attenuation and Transmittance through Filter Tape		No		
Sulfates	24 Hour	25 μg/m³	Ion Chromatography	National			
Hydrogen Sulfide	1 Hour	0.03 ppm (42 μg/m³)	Ultraviolet Fluorescence		Standards		

Gas

Chromatography

See footnotes on next page ...

Vinyl

Chloride¹²

 $0.01 \text{ ppm } (26 \text{ } \mu\text{g/m}^3)$

24 Hour

- 1. California standards for ozone, carbon monoxide (except 8-hour Lake Tahoe), sulfur dioxide (1 and 24 hour), nitrogen dioxide, and particulate matter (PM10, PM2.5, and visibility reducing particles), are values that are not to be exceeded. All others are not to be equaled or exceeded. California ambient air quality standards are listed in the Table of Standards in Section 70200 of Title 17 of the California Code of Regulations.
- 2. National standards (other than ozone, particulate matter, and those based on annual arithmetic mean) are not to be exceeded more than once a year. The ozone standard is attained when the fourth highest 8-hour concentration measured at each site in a year, averaged over three years, is equal to or less than the standard. For PM10, the 24 hour standard is attained when the expected number of days per calendar year with a 24-hour average concentration above 150 μg/m³ is equal to or less than one. For PM2.5, the 24 hour standard is attained when 98 percent of the daily concentrations, averaged over three years, are equal to or less than the standard. Contact the U.S. EPA for further clarification and current national policies.
- 3. Concentration expressed first in units in which it was promulgated. Equivalent units given in parentheses are based upon a reference temperature of 25°C and a reference pressure of 760 torr. Most measurements of air quality are to be corrected to a reference temperature of 25°C and a reference pressure of 760 torr; ppm in this table refers to ppm by volume, or micromoles of pollutant per mole of gas.
- 4. Any equivalent measurement method which can be shown to the satisfaction of the ARB to give equivalent results at or near the level of the air quality standard may be used.
- 5. National Primary Standards: The levels of air quality necessary, with an adequate margin of safety to protect the public health.
- 6. National Secondary Standards: The levels of air quality necessary to protect the public welfare from any known or anticipated adverse effects of a pollutant.
- 7. Reference method as described by the U.S. EPA. An "equivalent method" of measurement may be used but must have a "consistent relationship to the reference method" and must be approved by the U.S. EPA.
- 8. On October 1, 2015, the national 8-hour ozone primary and secondary standards were lowered from 0.075 to 0.070 ppm.
- 9. On February 7, 2024, the national annual PM2.5 primary standard was lowered from 12.0 μg/m³ to 9.0 μg/m³. The existing national 24-hour PM2.5 standards (primary and secondary) were retained at 35 μg/m³, as was the annual secondary standard of 15.0 μg/m³. The existing 24-hour PM10 standards (primary and secondary) of 150 μg/m³ also were retained. The form of the annual primary and secondary standards is the annual mean, averaged over 3 years.
- 10. To attain the 1-hour national standard, the 3-year average of the annual 98th percentile of the 1-hour daily maximum concentrations at each site must not exceed 100 ppb. Note that the national 1-hour standard is in units of parts per billion (ppb). California standards are in units of parts per million (ppm). To directly compare the national 1-hour standard to the California standards the units can be converted from ppb to ppm. In this case, the national standard of 100 ppb is identical to 0.100 ppm.
- 11. On June 2, 2010, a new 1-hour SO₂ standard was established and the existing 24-hour and annual primary standards were revoked. To attain the 1-hour national standard, the 3-year average of the annual 99th percentile of the 1-hour daily maximum concentrations at each site must not exceed 75 ppb. The 1971 SO₂ national standards (24-hour and annual) remain in effect until one year after an area is designated for the 2010 standard, except that in areas designated nonattainment for the 1971 standards, the 1971 standards remain in effect until implementation plans to attain or maintain the 2010 standards are approved.

 Note that the 1-hour national standard is in units of parts per billion (ppb). California standards are in units of parts per million (ppm). To directly compare the 1-hour national standard to the California standard the units can be converted to ppm. In this case, the national standard of 75 ppb is identical to 0.075 ppm.
- 12. The ARB has identified lead and vinyl chloride as 'toxic air contaminants' with no threshold level of exposure for adverse health effects determined. These actions allow for the implementation of control measures at levels below the ambient concentrations specified for these pollutants.
- 13. The national standard for lead was revised on October 15, 2008 to a rolling 3-month average. The 1978 lead standard (1.5 μg/m³ as a quarterly average) remains in effect until one year after an area is designated for the 2008 standard, except that in areas designated nonattainment for the 1978 standard, the 1978 standard remains in effect until implementation plans to attain or maintain the 2008 standard are approved.
- 14. In 1989, the ARB converted both the general statewide 10-mile visibility standard and the Lake Tahoe 30-mile visibility standard to instrumental equivalents, which are "extinction of 0.23 per kilometer" and "extinction of 0.07 per kilometer" for the statewide and Lake Tahoe Air Basin standards, respectively.

Thomas Jordan

From:

Kaelen McCrane <kaelen.mccrane@mainspringenergy.com>

Sent:

Wednesday, October 23, 2024 4:05 PM

To:

Thomas Jordan

Subject:

Re: Chemical release Data Planning Commission Mtg and need to address public

concerns

Attachments:

SVEC Slides 10.23.24.pptx; 230kW Linear Generator SCAQMD BACT Source Test

2.2.24.pdf; MSE3-250 Datasheet - R30203.pdf

CAUTION: This email originated from outside of the organization. Do not click links or open attachments unless you recognize the sender and know the content is safe.

Hi Tom,

Absolutely. I've been working with our environmental compliance and emissions team. The short version is that we expect the emissions profile to be very similar to that of our natural gas projects. I've attached a BACT (best available control technology) determination from South Coast Air Quality Management earlier this year which shows real source testing data on one of our units. This is public and ok to share. I've also included a few slides on Mainspring's emission profile vs other alternatives in case those are helpful.

The overall message is that because we're not burning any fuel, but creating precisely enough temperature and pressure to oxidize the fuel stream - Mainspring's linear generators create virtually zero criteria emissions regardless of the fuel composition. This system will be cleaner than any gas or diesel engine in Lake County.

We expect the syngas to be largely made up of Hydrogen, CO, CO2, and a small amount of Methane. This is an oversimplification but in case you're pressed, nearly all H2 becomes H2O or water vapor, CO becomes CO2, Methane becomes CO2 & H2O and the CO2 in the fuel does not react at all.

From an air permitting standpoint, we expect our emissions to be:

NOx <1.5ppm

CO <12ppm

VOC <10ppm

PM2.5 < .0019lbs/MMBtu

PM10 < .0025lbs/MMBtu

Hopefully this is sufficient, but let me know if you have any questions. Best of luck!

Kaelen McCrane

Schedule

a call with me



Mainspring